

Quantum Control of Photoelectron Circular Dichroism on Limonene

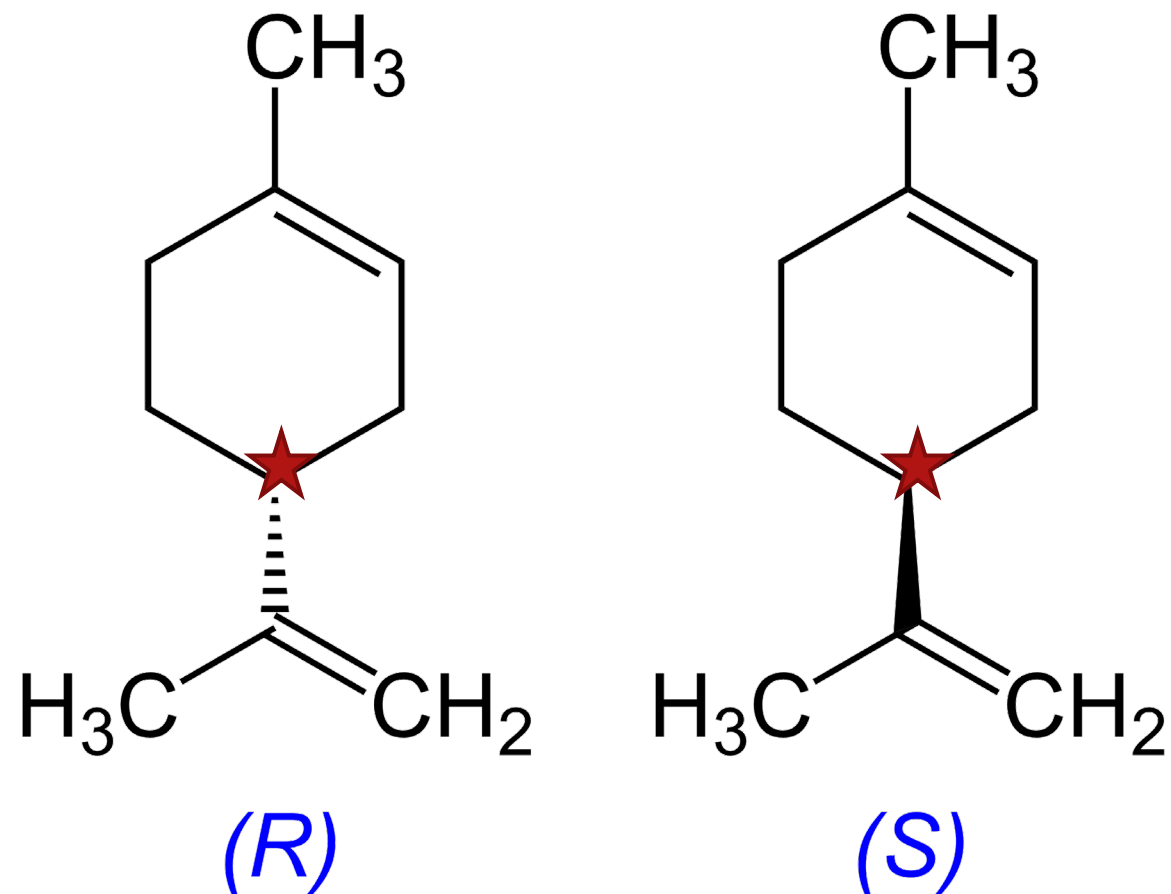
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K-STATE REU PROGRAM, AUGUST 26

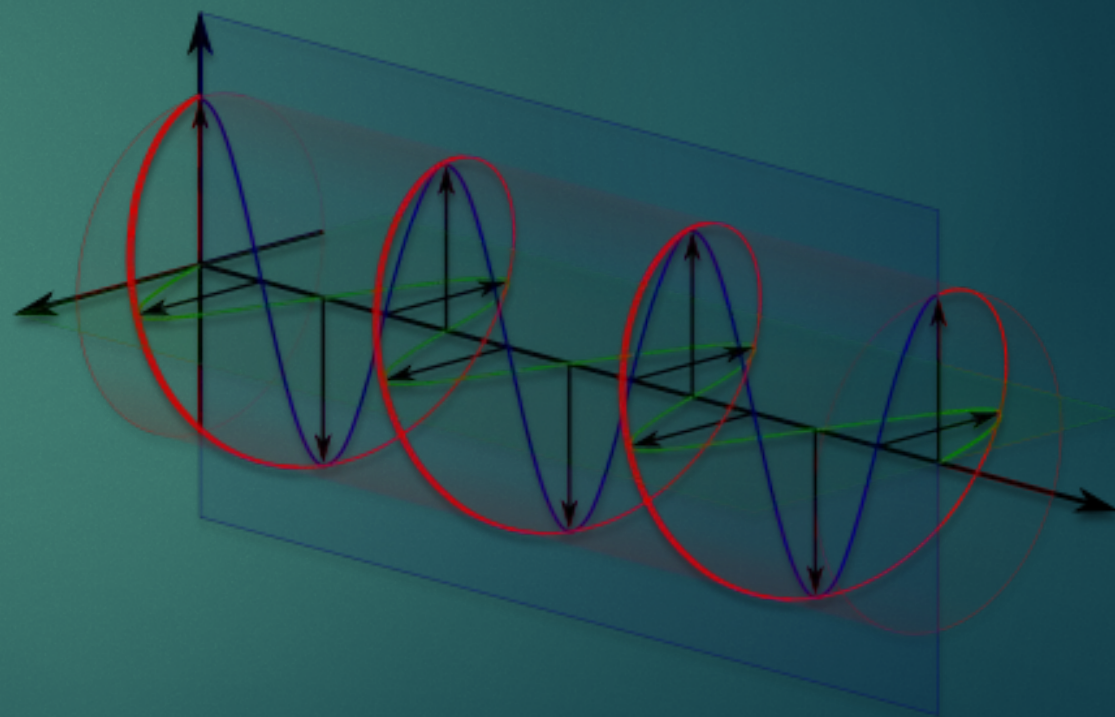
Limonene

- ▶ Found as major component of the oil in citrus peels
- ▶ It is a Chiral Molecule; non-superimposable with their mirror image
- ▶ Interesting due to its size and complexity



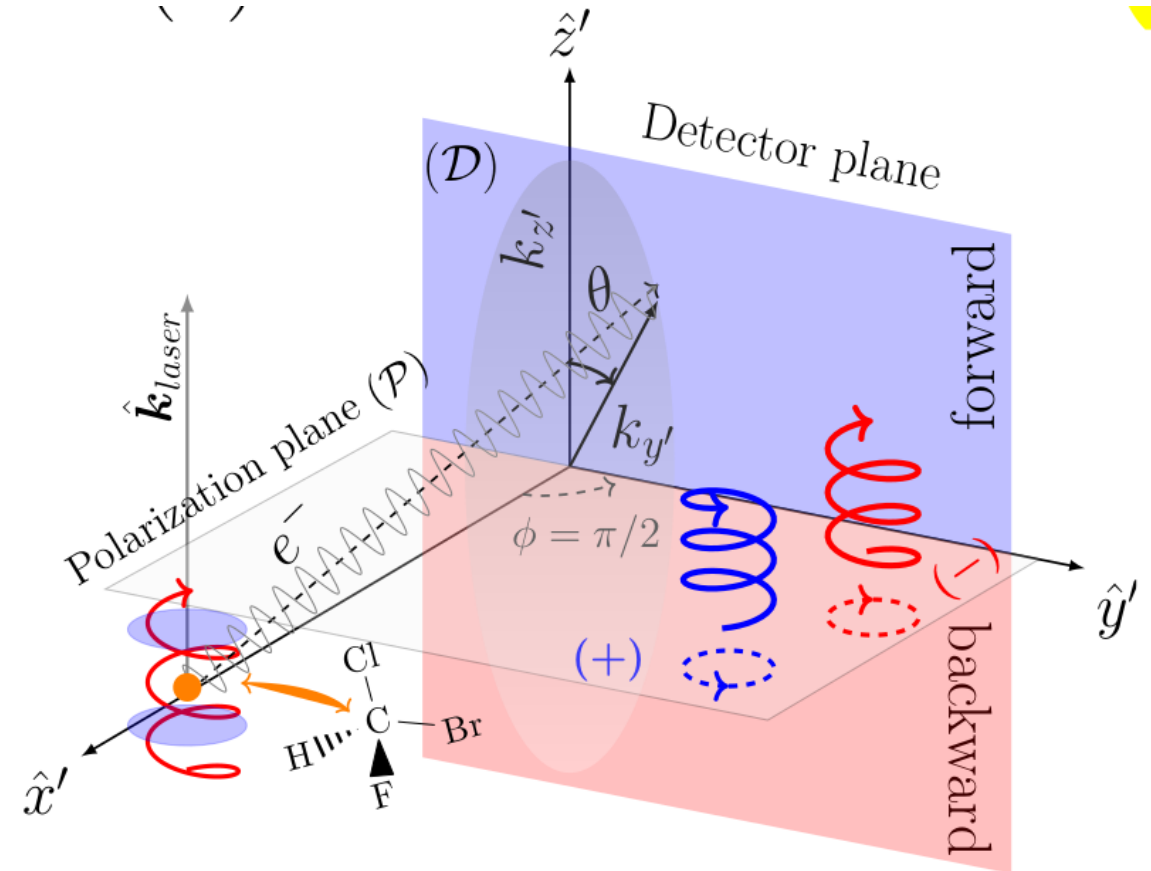
What is Circular Dichroism?

- ▶ Chiroptical effect
- ▶ Based on the differential absorption of left and right circularly polarized light
- ▶ Make use of magnetic dipole moment of the molecule



Photoelectron Circular Dichroism (PECD)

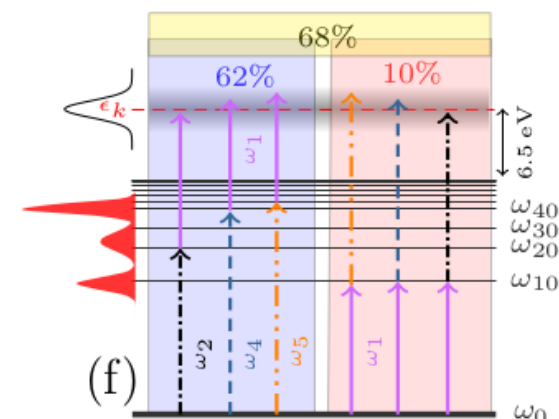
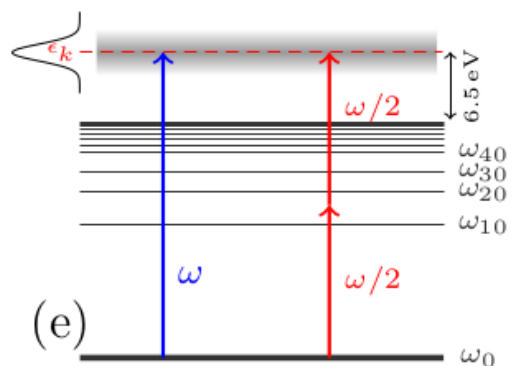
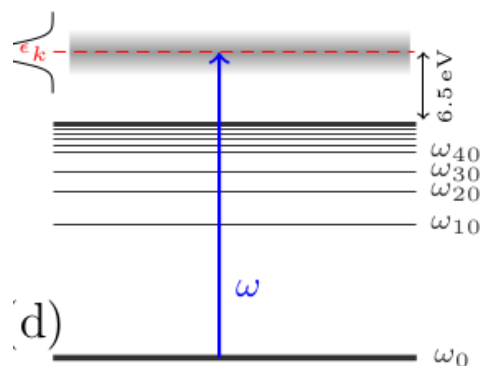
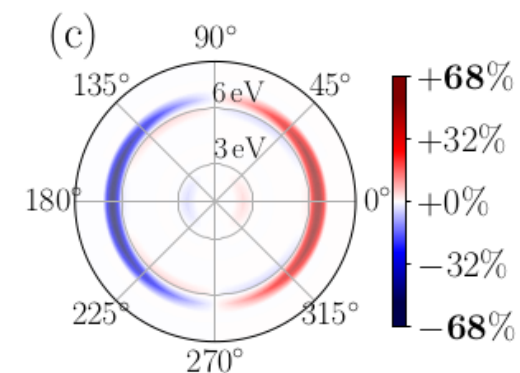
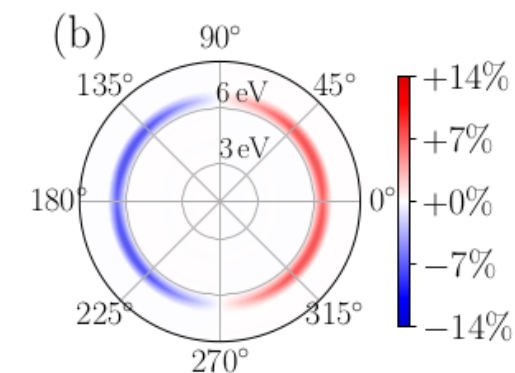
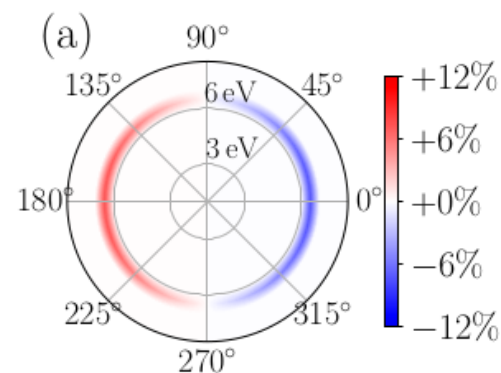
- ▶ Make use of Electric dipole moment
- ▶ Much stronger signal than CD
- ▶ Sensitive to
 - Electronic Structure
 - Vibrational Excitation
 - Molecular Conformation
- ▶ Differential in photo-electron angular distribution



What's Been Done Before

► Angular resolved PECD

- One-Photon ionization
- Bichromatic pulse
- Two-photon pathway



Structure of Research

- ▶ Time-independent description of Limonene
 - ▶ Description of Limonene orbitals
 - ▶ Description of photoionization
- ▶ Time dependent behavior
 - ▶ Interaction with laser pulse

Obtaining Limonene Orbitals

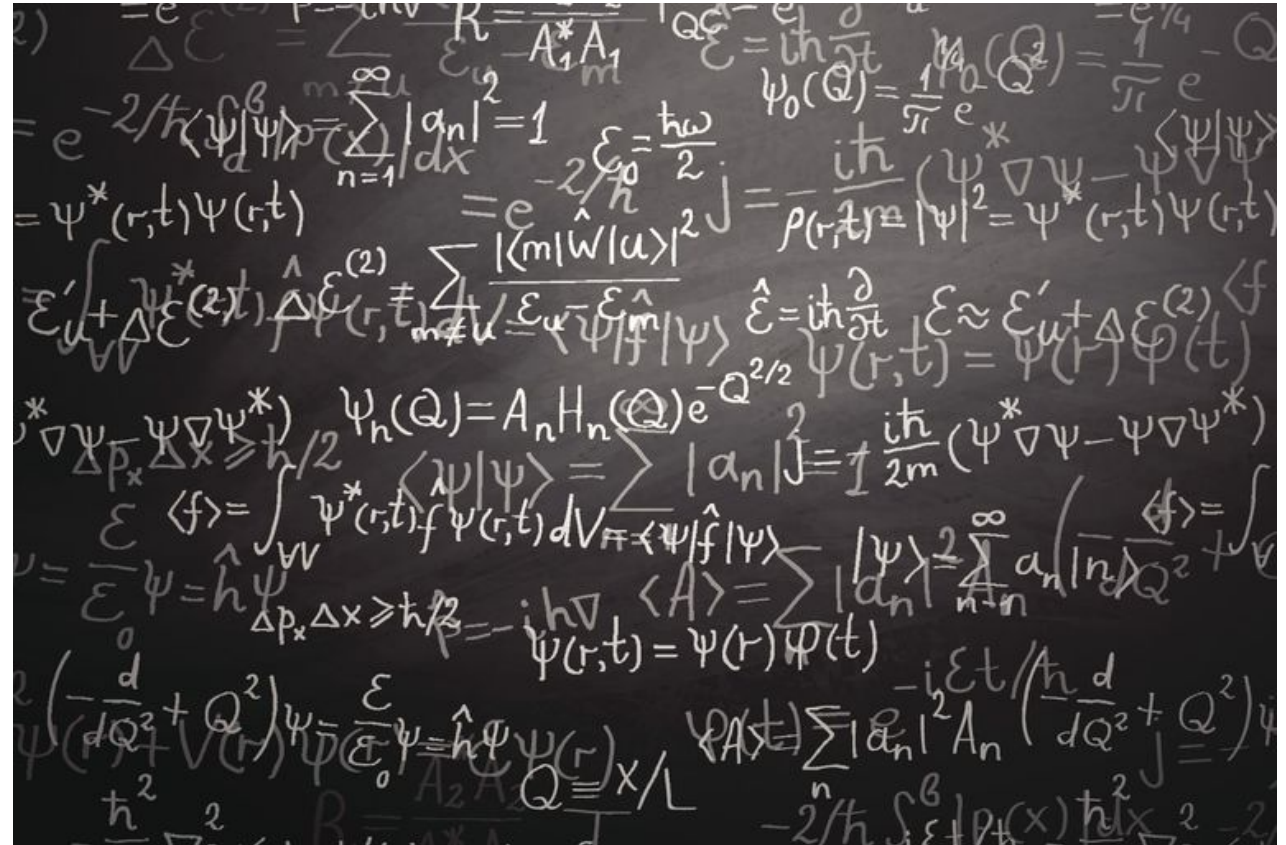
- ▶ Schrödinger Equation

$$-\frac{\hbar^2}{2m}\nabla^2\psi + V(\mathbf{x})\psi = E\psi$$

- ▶ Hartree-Fock Approximation

- ▶ Make a multi-electron system into a single electron system

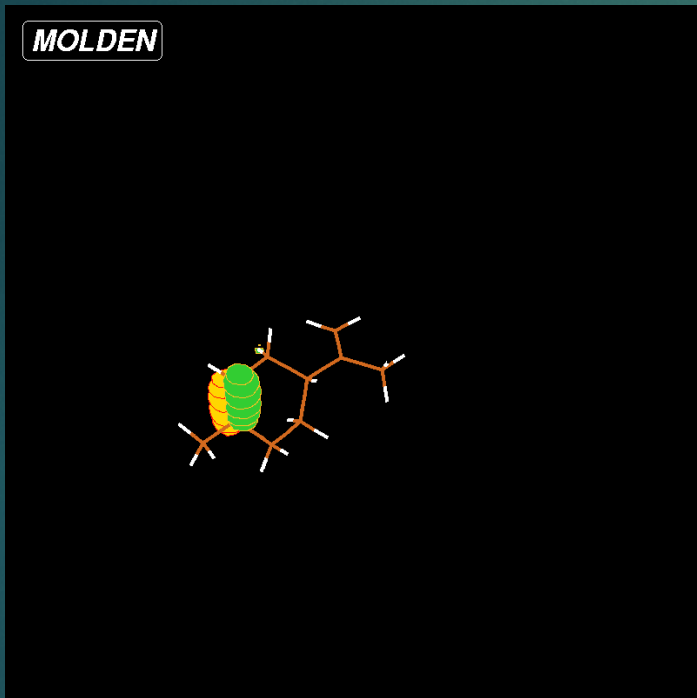
- ▶ Solved using Molpro in NERSC's Cori supercomputer



Limonene Orbitals

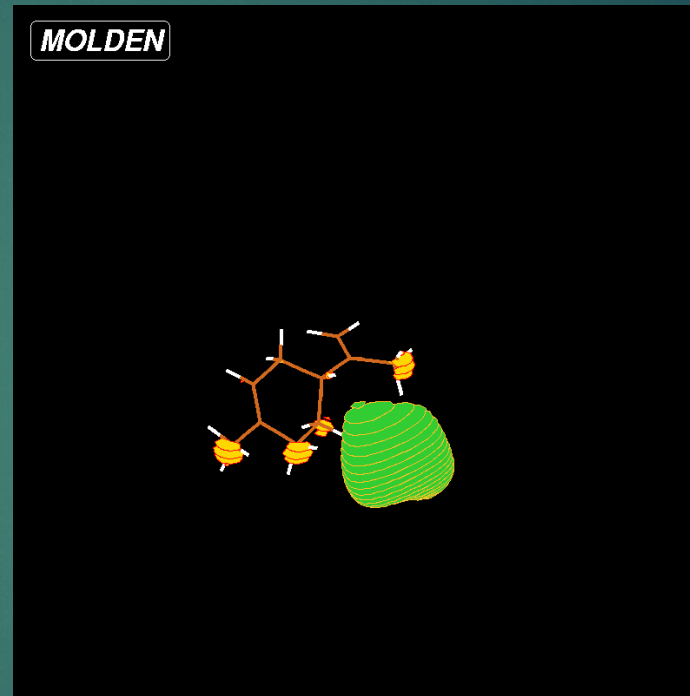
Highest Occupied Molecular Level

$$\epsilon = -9.0 \text{ eV}$$



Lowest Unoccupied Molecular Level

$$\epsilon = 0.95 \text{ eV}$$



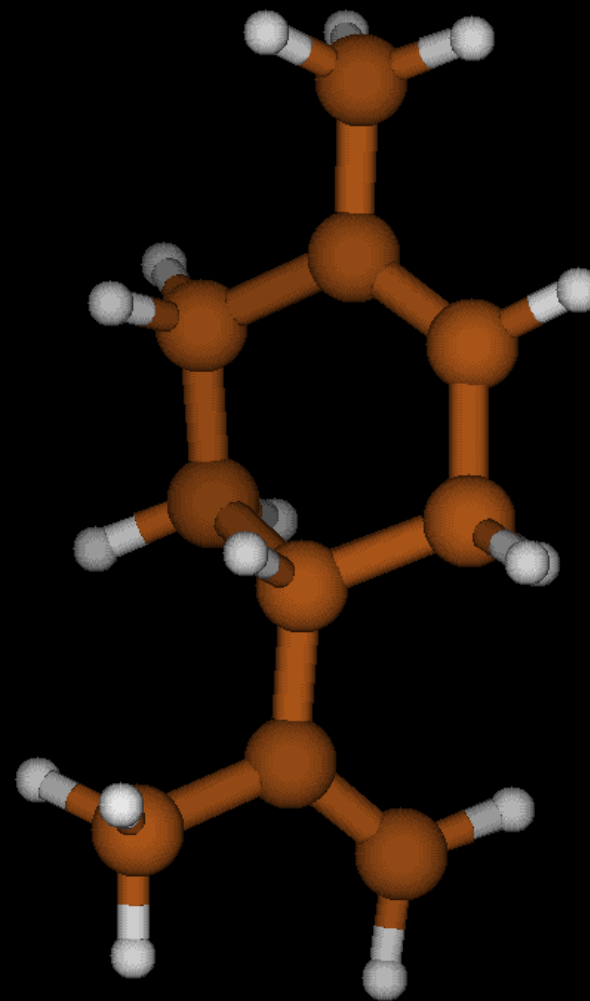
Experimental Ionization Energy = 8.3 eV

Calculated Ionization Energy = 9.98 eV

Photoionization

- ▶ Simulated using ePolyScat; software used to study photoelectron scattering
- ▶ Produces the photoionization dipole matrix elements at a given energy.

MOLDEN



Outlook & Conclusion

- ▶ Finish time-independent description of Limonene
- ▶ Move into quantum control of PECD for Limonene
- ▶ Compare to the results found on CHBrClF
- ▶ Expand the tools we have for probing chiral molecules

Acknowledgments

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- ▶ Dr. Bret Flanders
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- ▶ Kansas State University
- ▶ National Science Foundation



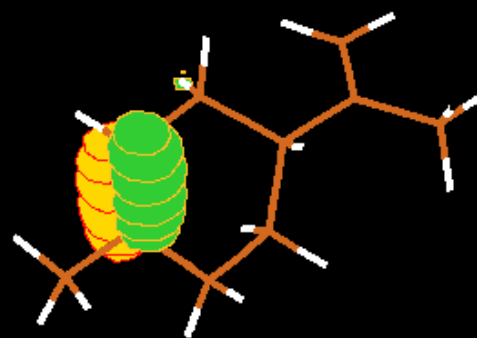
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Limonene Orbitals

- ▶ Limonene
HOMO with
aug-cc-pVDZ
- ▶ Energy HOMO
= -9.0446 eV

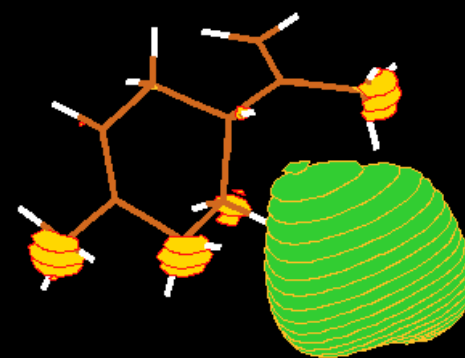
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Limonene Orbitals

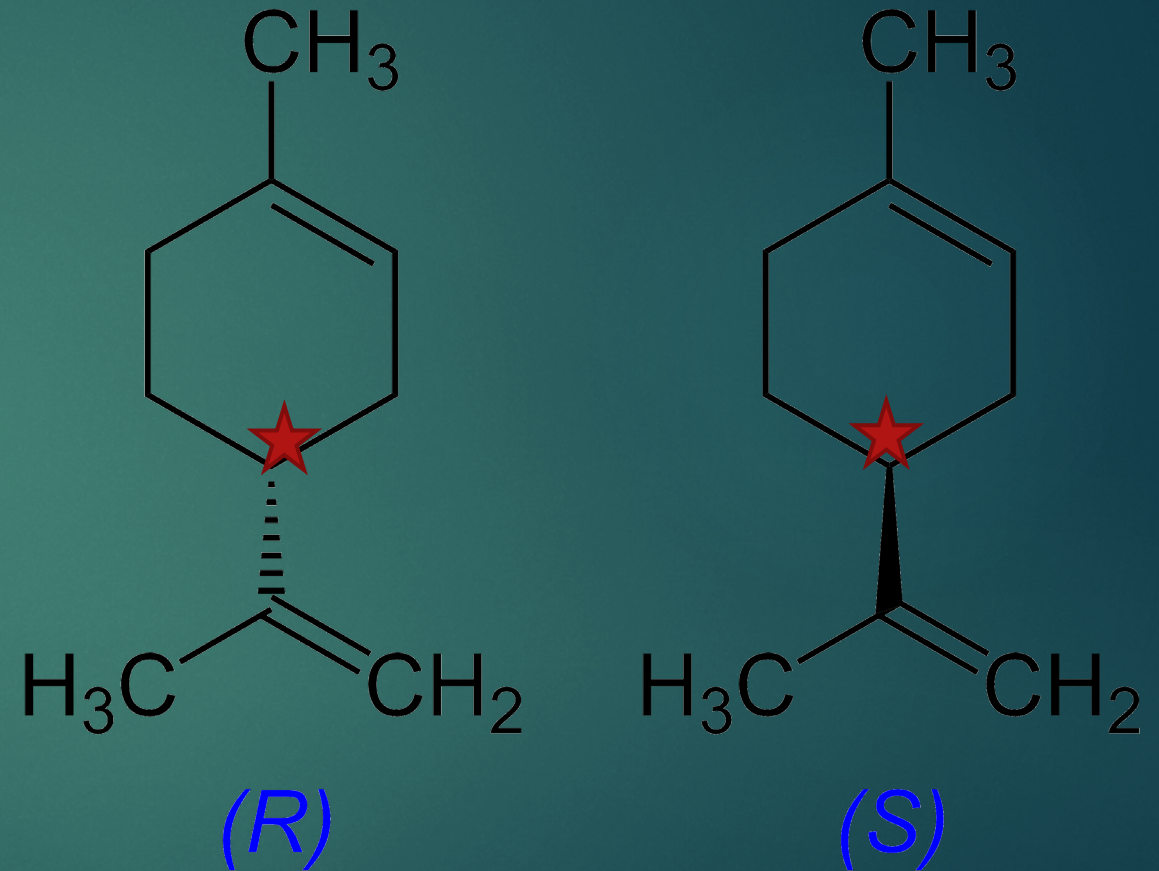
- ▶ Limonene LUMO
with aug-cc-
pVDZ
- ▶ Energy LUMO =
0.7729 eV

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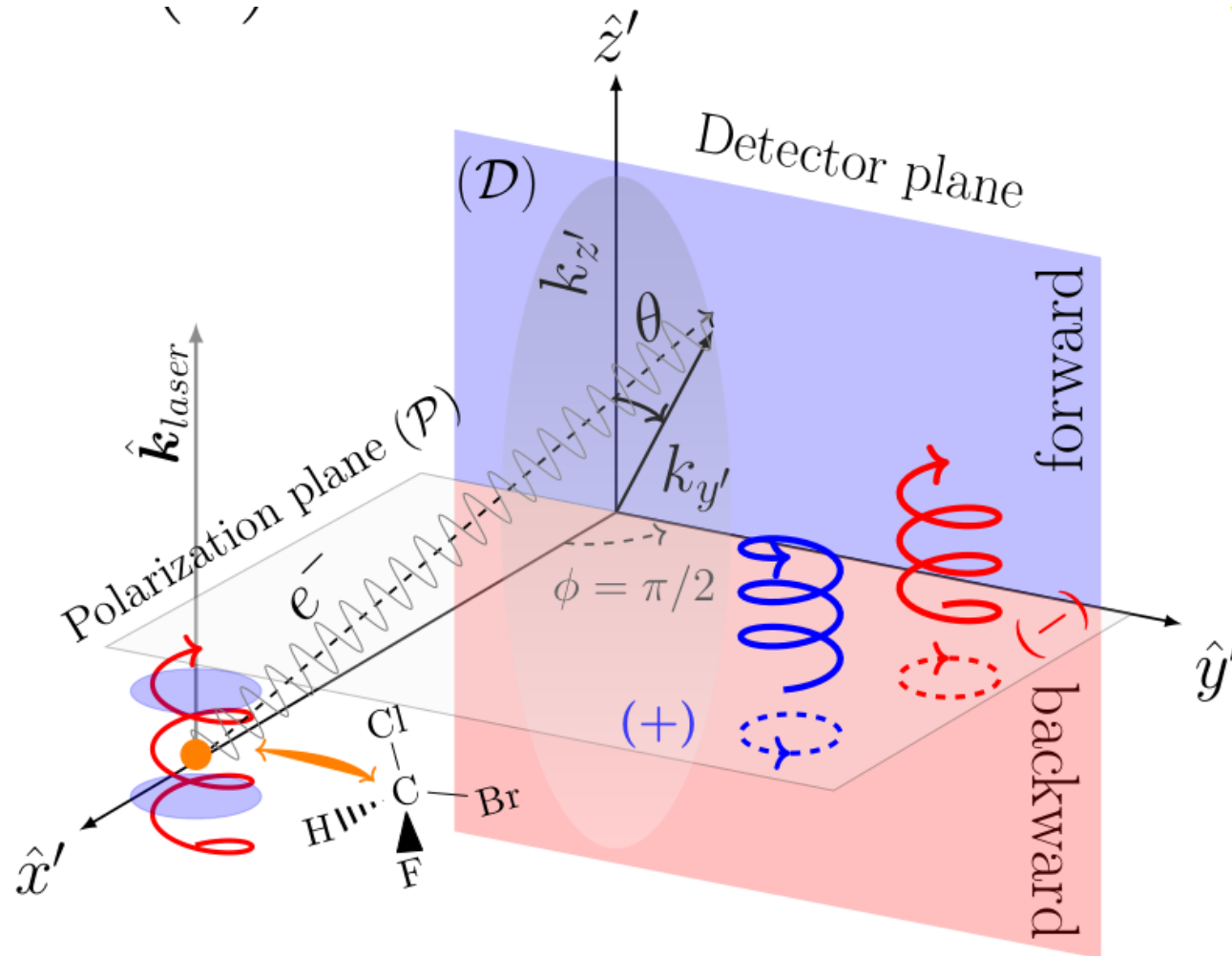


Limonene

- ▶ Found as major component of the oil in citrus peels
- ▶ It is a Chiral Molecule; non-superimposable with their mirror image
- ▶ Uses
 - ▶ Dietary supplements, Fragrance, Ingredients for cosmetics
 - ▶ Solvent for fused filament fabrication-based 3D printing



- Worked with CHBrCIF



Case Study